

Organic Chemistry Education

My work in organic chemistry education has produced (1) hand-held devices to facilitate learning undergraduate organic and general chemistry [5-11], (2) evaluating established education protocols for undergraduate chemistry students and developing novel improvements [1-3,14], and (3) interesting ways to present research to the public [4,13]. Each of these is discussed below.

Organic chemistry is often described as a difficult discipline because there is such a large amount of information to be learned. There is a need to visualize organic reaction pathways and become familiar with the key steps in many reaction mechanisms. In order to facilitate pattern recognition and learning the reactions rather than memorizing them, we developed three teaching devices for undergraduate organic chemistry. Two were adopted by publishers for use in conjunction with organic chemistry texts [6]. The three teaching devices for undergraduate organic chemistry are: (1) the Nucleophile / Electrophile (Nu/E) Reaction Guide for the reaction of electrophiles with nucleophiles [5,6,9-11], available commercially, [5,6] (2) the Electrophilic Aromatic Substitution (EAS) Tool for learning products, reagent generation, substituent effects, and mechanisms in electrophilic aromatic substitution [7,8,12], and (3) Reaction Site Selection in Carbonyl Compounds device (Carbonyl Site Selector) with three techniques, and the rationales, for predicting the site of attack by a nucleophile or base. We have studied the effect of each of these three devices on student test performance. We found that the Nu/E Guide improved the pertinent test question performance of students using the device outside of class by 20.1% over those students who did not use the device at all, by 9.1% over those students who used the device only in class and by 8.1% over those students who used the device both inside and outside of class. We surveyed students further and gathered information which led to the conclusion that students using the device inside class were distracted from explanations and discussion being presented, which reduced learning. The EAS Tool produced analogous improvements of 17.3%, 4.4%, and 3.8% respectively. Results for the Carbonyl Site Selector are being calculated.

The large size of the undergraduate organic chemistry lecture classes was of concern. This was exacerbated by lack of teaching assistants, who had passed University English exams and were therefore able to have contact with students. Some semesters, a TA was provided who could have no contact with students, but poor English rendered them unable to do more than wash transparencies and draw figures for presentation in class. Consequently exams were multiple choice and computer graded. The question how other regional universities were staffing these courses, prompted surveying organic chemistry professors in Big 12 schools and publishing the results.[14] One school other than ours did not provide lecture TAs who could have contact with students; that school was a private school with much smaller organic class sizes.

One research project, undertaken to increase the appeal of science to the public and in particular to middle and high school students, was recently selected as the subject of a permanent exhibit in the Oklahoma Science Museum (formerly Oklahoma City Air and Space Museum) [13]. (At the time, there were only four privately funded such air and space museums in the US, one being in the Smithsonian, which also considered making our project part of its Life Sciences exhibit, but eliminated it in the final cut.) We analyzed precipitates formed in a series of reactions carried out aboard the "Getaway Special" in a project designed specifically to inspire Oklahoma high school students' interest in science. The project description, the experiment materials, and the equipment, such as the space capsule holding the remotely-controlled reaction chambers, constitute an exhibit at the Oklahoma Science Museum. Appropriately, this space project which was conceived, sponsored, constructed, and analyzed all by Oklahomans will be displayed at the museum containing the contributions of many other Oklahomans. The exhibit has been seen by thousands of Oklahoma students each year; we hope that exposure to an experiment conceived, constructed, funded, and analyzed by other Oklahomans was carried out on board the Space Shuttle, will inspire future Oklahoma students into science as well. Dr. Nelson's part in this was (1) to conceive, design, and carry out sample analysis, (2) to prepare the text, samples, and analysis photos for the museum display, and (3) to have the research published in a refereed journal [4]. It was a challenge to

provide an explanation of the project which a 10-year-old would find interesting, comprehensible, and inspirational.

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